

# GEOLOGICAL AND MINERALOGICAL ASPECTS ON MINERAL CARBONATION OF ROCKS AND MINE TAILINGS

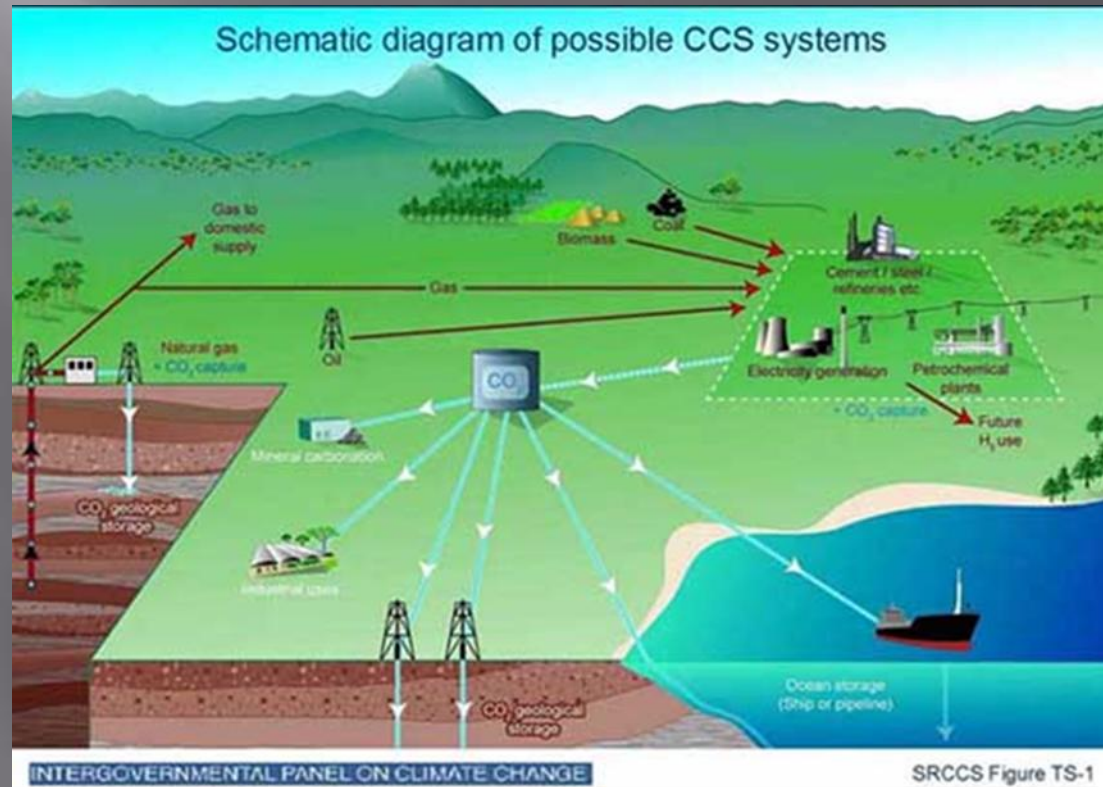


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Anthropogenic CO<sub>2</sub> emissions and climate change

Carbon Capture and Storage  
- CCS

Carbon Capture and Mineralization  
- CCM



# My research

Questions to be answered:

- 1: What are the abilities of different minerals and rocks to bind carbon dioxide?
- 2: How the origin of the rocks affects the tests?
- 3: Why some minerals have higher reactivity than others?
- 4: What circumstances produce the most cost-effective results?



# Materials

**Test materials:** Mine tailings / waste rocks

**Cooperating mines:** Kevitsa, Hitura, Pampalo, Horsmanaho (Mondo Minerals) and Talvivaara

**Main interest:** ultramafic rocks with low  $\text{SiO}_2$  and high MgO content

**Mg-rich host rocks provided :**

olivine  $(\text{Mg,Fe})_2\text{SiO}_4$

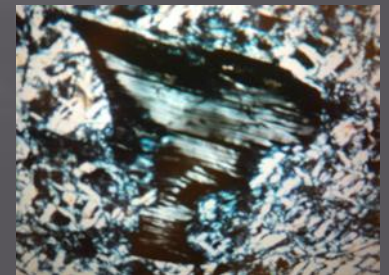
pyroxene  $\text{Ca}(\text{Mg,Fe})\text{Si}_2\text{O}_6$

serpentine  $(\text{Mg,Fe})_3\text{Si}_2\text{O}_5(\text{OH})_4$

talc  $\text{Mg}_3\text{Si}_4\text{O}_{10}(\text{OH})_2$

chlorite  $(\text{Mg}_5\text{Al})(\text{AlSi}_3)\text{O}_{10}(\text{OH})_8$

**The most promising rock type:** Serpentinite



# Methods

Mineralogy and structure (optic microscope)

Surface analyses (Scanning electron microscope, SEM)

Mineral phases (X-ray Diffraction, XRD)

Chemical composition (X-ray Fluorescence, XRF)

Bonding environments (Electron Spectroscopy for Chemical Analysis, ESCA)



# Examples of CCM in practice

**Information considering examples of CCM in practice can be found (e.g) in:**

Gunning PJ, Hills CD, Carey PJ 2009.

Production of lightweight aggregate from industrial waste and carbon dioxide.  
Waste Management 2009 Oct;29(10):2722-8

**More examples on green building can be found on:**

[www.lignacite.co.uk](http://www.lignacite.co.uk)

# Thank you!

(Choosing wisely gives us delicious wine and beer in the future)

